

# Two-step synthesis of *C*-glycosyl juglones from unprotected sugars: a novel approach to angucycline antibiotics

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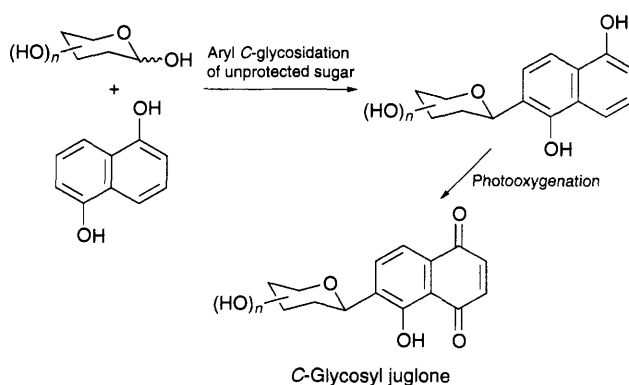
The two-step synthesis of *C*-glycosyl juglones, versatile synthetic intermediates for angucycline antibiotics, has been developed by the *C*-glycosidation of naphthalene-1,5-diol with an unprotected sugar and the subsequent regioselective photooxygenation of the resultant *C*-glycosyl naphthalenediol.

*C*-Glycosyl juglone, a 5-hydroxy-1,4-naphthoquinone bearing a *C*-glycoside at C6, is a common structural feature among the many members of angucycline antibiotics such as aquayamycin **1** and urdamycinone **B 2** which show a variety of biological activity including antitumour activity.<sup>1</sup> *C*-Glycosyl juglone is also a promising synthetic intermediate for this class of antibiotics. Therefore, several approaches to *C*-glycosyl juglone have been developed<sup>2</sup> and the syntheses of angucycline antibiotics *via* the *C*-glycosyl juglone have also been reported.<sup>3,4</sup> Herein we report a novel and efficient approach to *C*-glycosyl juglones in two steps involving the *C*-glycosidation of naphthalene-1,5-diol with an unprotected sugar to form the aryl *C*-glycosidic linkage and the subsequent regioselective photooxygenation of the resultant *C*-glycosyl naphthalenediol to construct the juglone skeleton (Scheme 1). To the best of our knowledge, this is the first protocol for the synthesis of *C*-glycosyl juglones without any protecting groups on the aglycon or glycon moieties.

The first step is the *C*-glycosidations<sup>5</sup> of naphthalene-1,5-diol **3** with several unprotected sugars **4–7**. The results are summarized in Table 1. D-Olucose **4** is the most representative sugar that appeared in many angucycline antibiotics. The *C*-glycosidation of **3** (2.0 equiv.) and **4** (1.0 equiv.) using trimethylsilyl trifluoromethanesulfonate (Me<sub>3</sub>SiOTf) (0.2 equiv.)<sup>6</sup> in MeCN at 25 °C for 1 h proceeded smoothly to give the unprotected aryl β-*C*-glycoside **8**<sup>†</sup> in 65% yield as a single isomer (entry 1 in Table 1). Unprotected methyl D-olivoside **5** was also coupled with **3** under similar conditions to afford **8** in good yield (entry 2 in Table 1). To enhance the synthetic utility of this reaction, we further examined the *C*-glycosidations of other unprotected 2,6-dideoxy sugars **6** and **7** with **3** mediated

by a catalytic amount of Me<sub>3</sub>SiOTf. It was found that these *C*-glycosidations also proceeded easily to furnish the unprotected aryl β-*C*-glycosides **9**<sup>†</sup> and **10**<sup>†</sup> in 61 and 65% yields, respectively (entries 3 and 4 in Table 1).

With the unprotected *C*-glycosyl naphthalenediols in hand, the second step, the oxygenation to the *C*-glycosyl juglones, was carried out. We first examined the oxygenation of **8** using ammonium cerium(IV) nitrate (CAN),<sup>7</sup> potassium nitrosodisulfonate (Fremy's salt)<sup>8</sup> and thallium(III) nitrate trihydrate (TTN)<sup>9</sup> as the oxygenating agents, which are widely used for



Scheme 1

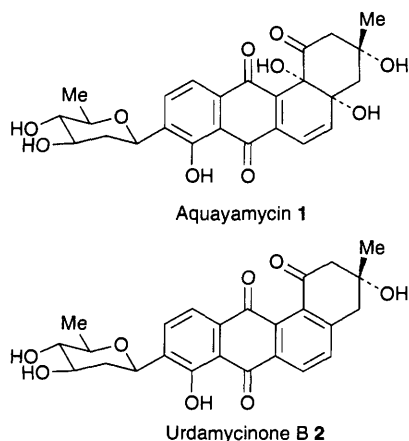
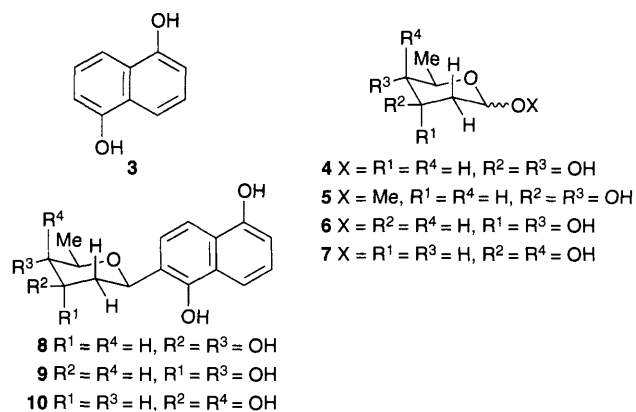
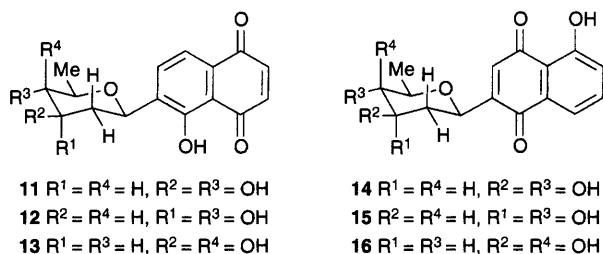


Table 1 *C*-Glycosidations of **3** and the unprotected sugars **4–7** to give the β-*C*-glycosides **8–10**<sup>a</sup>

Entry	Sugar	Product	Yield (%)
1	<b>4</b>	<b>8</b>	65
2	<b>5</b>	<b>8</b>	64
3	<b>6</b>	<b>9</b>	61
4	<b>7</b>	<b>10</b>	65

<sup>a</sup> All reactions were carried out using 2.0 equiv. of **3** with respect to the sugars.

conversion of naphthol to quinone. However, in these cases, the desired *C*-glycosyl juglone **11**† was either not detected at all or isolated in a very low yield while the regioisomer **14**‡ was predominantly produced as shown in Table 2. Therefore, our attention next turned to the photooxygenation<sup>10</sup> of **8**. These results are summarized in Table 3. Remarkably, the regioselectivity of the oxygenation of **8** was dramatically changed and the desired *C*-glycosyl juglone **11** was predominantly obtained by choice of the appropriate reaction solvent. Thus, the photooxygenation of **8**, without any reagent, was best effected by irradiating it with sunlight in Bu<sup>t</sup>OH–CHCl<sub>3</sub> (1 : 3, 0.0069 M



**Table 2** Oxygenations of **8** at 25 °C to give the *C*-glycosyl juglones **11** and **14**

Entry	Reagent (equiv.)	Solvent	t/h	Yield (%)	
				<b>11</b>	<b>14</b>
1	CAN (2.5)	MeCN–H <sub>2</sub> O	0.5	0	25
2	Fremy's salt (3.6)	MeOH	1	10	55
3	TTN (1.8)	MeOH	0.5	1	74

**Table 3** Photooxygenations of **8**–**10** to give the *C*-glycosyl juglones **11**–**16**

Entry	<i>C</i> -Glycosyl naphthalenediol	Solvent <sup>a</sup>	<i>C</i> -Glycosyl juglone/ Yield (%)			
			<b>11</b>	<b>7</b>	<b>14</b>	<b>12</b>
1	<b>8</b>	MeOH	<b>11</b>	<b>7</b>	<b>14</b>	<b>12</b>
2	<b>8</b>	EtOH	<b>11</b>	<b>19</b>	<b>14</b>	<b>17</b>
3	<b>8</b>	Pr <sup>i</sup> OH	<b>11</b>	<b>30</b>	<b>14</b>	<b>17</b>
4	<b>8</b>	Bu <sup>t</sup> OH	<b>11</b>	<b>52</b>	<b>14</b>	<b>22</b>
5	<b>8</b>	Bu <sup>t</sup> OH–hexane(1 : 1)	<b>11</b>	<b>41</b>	<b>14</b>	<b>13</b>
6	<b>8</b>	Bu <sup>t</sup> OH–PhH(1 : 1)	<b>11</b>	<b>51</b>	<b>14</b>	<b>15</b>
7	<b>8</b>	Bu <sup>t</sup> OH–CHCl <sub>3</sub> (1 : 1)	<b>11</b>	<b>52</b>	<b>14</b>	<b>14</b>
8	<b>8</b>	Bu <sup>t</sup> OH–CHCl <sub>3</sub> (1 : 3)	<b>11</b>	<b>57</b>	<b>14</b>	<b>13</b>
9	<b>9</b>	Bu <sup>t</sup> OH–CHCl <sub>3</sub> (1 : 1)	<b>12</b>	<b>56</b>	<b>15</b>	<b>17</b>
10	<b>10</b>	Bu <sup>t</sup> OH–CHCl <sub>3</sub> (1 : 1)	<b>13</b>	<b>55</b>	<b>16</b>	<b>18</b>

<sup>a</sup> 0.0069 M for *C*-glycosyl naphthalenediol.

for **8**) under O<sub>2</sub> at room temperature for 12 h to give the *C*-glycosyl juglone **11** in 57% yield along with a 13% yield of **14** (entry 8 in Table 3). Other *C*-glycosyl naphthalenediols **9** and **10** were also found to cleanly convert under similar conditions to the *C*-glycosyl juglones **12** and **13** in 56 and 55% yields, respectively (entries 9 and 10 in Table 3).

A typical experimental procedure for the preparation of **8** via *C*-glycosidation of **3** and **4**, followed by photooxygenation to give **11**, is as follows. To a mixture of **3** (183.8 mg, 1.148 mmol) and **4** (85.0 mg, 0.574 mmol) in dry MeCN (5.74 ml) was added Me<sub>3</sub>SiOTf (0.022 ml, 0.115 mmol) under ice-cooling conditions. After stirring for 1 h at 25 °C, the reaction was quenched with Et<sub>3</sub>N and the resulting mixture was then concentrated *in vacuo*. Purification of the residue by flash column chromatography with 10 : 1 chloroform–methanol gave **8** (108.8 mg, 65.3%, β only) as a white solid. A solution of **8** (108.8 mg, 0.375 mmol) in Bu<sup>t</sup>OH–CHCl<sub>3</sub> (3 : 1, 54.4 ml) was then irradiated with diffuse sunlight (a 75 W xenon lamp, Wacom Sunray Lamp, I-Sunsun) under O<sub>2</sub> for 12 h and then concentrated *in vacuo*. Purification of the residue by flash column chromatography with 2 : 1 hexane–acetone gave **11** (64.4 mg, 56.5%) and **14** (14.8 mg, 13.0%) as an orange solid.

## Footnotes

† The β-configuration of *C*-glycoside was evident from the <sup>1</sup>H NMR spectra.

‡ Structure confirmation relies on the NOE experiments.

## References

- J. Rohr and R. Thiericke, *Nat. Prod. Rep.*, 1992, 103.
- F. L. Andrews and D. S. Larsen, *Tetrahedron Lett.*, 1994, **35**, 8693; T. Matsumoto, T. Sohma, H. Yamaguchi and K. Suzuki, *Chem. Lett.*, 1995, 677.
- V. A. Boyd and G. A. Sulikowski, *J. Am. Chem. Soc.*, 1995, **117**, 8472.
- G. Matsuo, Y. Miki, M. Nakata, S. Matsumura and K. Toshima, *Chem. Commun.*, 1996, 225.
- M. H. D. Postema, *Tetrahedron*, 1992, **40**, 8545; D. E. Levy and C. Tang, *The Chemistry of C-Glycosides*, Pergamon Press, Oxford, 1995.
- K. Toshima, G. Matsuo, T. Ishizuka, M. Nakata and M. Kinoshita, *J. Chem. Soc., Chem. Commun.*, 1992, 1641; K. Toshima, G. Matsuo and M. Nakata, *J. Chem. Soc., Chem. Commun.*, 1994, 997.
- M. P. Sibi, J. W. Dankwardt and V. Snieckus, *J. Org. Chem.*, 1986, **51**, 271.
- H.-J. Teuber and N. Götz, *Chem. Ber.*, 1954, **87**, 1236.
- D. J. Crouse, M. M. Wheeler, M. Goemann, P. S. Tobin, S. K. Basu and D. M. S. Wheeler, *J. Org. Chem.*, 1981, **46**, 1814.
- K. Pfoertner and D. Böse, *Helv. Chim. Acta*, 1970, **53**, 1553.

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